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support of the preservation process, and thank you for being an important part of keeping this knowledge alive and relevant. Excerpt from Electrochemical Analysis: Studies of Acids, Bases, and Salts by Emf, Conductance, Optical and Kinetic Methods, July 1964 to June 1965 The first goal could be achieved in a most satisfactory way by collecting together the published or soon to be published work of the staff as listed at the end of this document. The second aim, however, is more elusive. It can only be met by an integrated summary of the total Section effort, where accomplishment can be viewed against the

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significance""are investigated and explained. Acids, bases and salts (chemlab) Excerpt from Electrochemical Analysis: Studies of Acids, Bases, and Salts by Emf, Conductance, Optical, and Kinetic Methods; July 1965 to June 1966 This is the second in a series of annual progress reports of the Electrochemical Analysis Section of the Analytical Chemistry Division. The report covers the fiscal year 1966, which began on July 1, 1965, and ended on June 30, 1966. Many of the processes and reactions of analytical interest take place in solutions, and a large fraction of these involve ionized solutes. If the research programs of the

Electrochemical Analysis Section were to be placed in a single broad category, undoubtedly Solution Electro chemistry would be a fair choice, with primary emphasis on acid - base phenomena, solvent effects on the behavior of electrolytes, and potentiometry with reversible electrodes. Competence in polarography and coulometry exists elsewhere in the Analytical Chemistry Division; hence, these areas are not a part of the research activity of the Electro chemical Analysis Section. In line with a uniform policy of the Division, the Section's programs have both research and sample aspects. During the fiscal year Just

ending, about 70 percent of the total effort was devoted to research, while 20 percent was devoted to programs on Standard Reference Materials and 10 percent to other-agency programs. The outstanding event of the present year was the long awaited move to the excellent new facility at Gaithersburg, Md. The move and the attendant loss of time during re settlement have inevitably left their mark on the Section's activity. More serious, however, has been a shortage of personnel. Two project leaders, Dr. Robert Gary and Dr. Richard K. Wolford, were chosen as Science and Technology Fellows and were assigned

elsewhere in the Department of Commerce for 10 months of the reporting period. A third, Dr. Marion M. Davis, retired from the Section on December 31, 1965. On the other hand, Dr. Paul w. Schindler spent nine months in the Section as a guest worker supported by the Swiss National Foundation. The purpose of this report is to summarize the broad program of the Electrochemical Analysis Section and to convey also the manner in which the individual projects contribute to the whole. An attempt is made to set forth in a rather complete way the entire year's activity of the Section and to reveal the ways in which this specialized group contributes to the

missions of the Division and Institute of which it is a part. About the Publisher Forgotten Books publishes hundreds of thousands of rare and classic books. Find more at www.forgottenbooks.com This book is a reproduction of an important historical work. Forgotten Books uses state-of-the-art technology to digitally reconstruct the work, preserving the original format whilst repairing imperfections present in the aged copy. In rare cases, an imperfection in the original, such as a blemish or missing page, may be replicated in our edition. We do, however, repair the vast majority of imperfections successfully; any imperfections

that remain are intentionally left to preserve the state of such historical works. A method for conducting a chemical reaction in acidic, basic, or neutral solution as required and then regenerating the acid, base, or salt by means of ion exchange in a closed

cycle reaction sequence which comprises contacting the spent acid, base, or salt with an ion exchanger, preferably a synthetic organic ion-exchange resin, so selected that the counter ions thereof are ions also produced as a by-product

in the closed reaction cycle, and then regenerating the spent ion exchanger by contact with the by-product counter ions. The method is particularly applicable to closed cycle processes for the thermochemical production of hydrogen.